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# NaCl-saturated brines are thermodynamically moderate, rather than extreme, microbial habitats

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#### **Abstract**

NaCl-saturated brines such as saltern crystalliser ponds, inland salt lakes, deepsea brines and liquids-of-deliquescence on halite are commonly regarded as a paradigm for the limit of life on Earth. There are, however, other habitats that are thermodynamically more extreme. Typically, NaCl-saturated environments contain all domains of life and perform complete biogeochemical cycling. Despite their reduced water activity, ~0.755 at 5 M NaCl, some halophiles belonging to the *Archaea* and *Bacteria* exhibit optimum growth/metabolism in these brines.

Furthermore, the recognised water-activity limit for microbial function,  $\sim$ 0.585 for some strains of fungi, lies far below 0.755. Other biophysical constraints on the microbial biosphere (temperatures of >121°C; pH > 12; and high chaotropicity; e.g. ethanol at >18.9% w/v (24% v/v) and MgCl<sub>2</sub> at >3.03 M) can prevent any cellular metabolism or ecosystem function. By contrast, NaCl-saturated environments contain biomass-dense, metabolically diverse, highly active and complex microbial ecosystems; and this underscores their moderate character. Here, we survey the evidence that NaCl-saturated brines are biologically permissive, fertile habitats that are thermodynamically mid-range rather than extreme. Indeed, were NaCl sufficiently soluble, some halophiles might grow at concentrations of up to 8 M. It may be that the finite solubility of NaCl has stabilised the genetic composition of halophile populations and limited the action of natural selection in driving halophile evolution towards greater xerophilicity. Further implications are considered for the origin(s) of life and other aspects of astrobiology.

Keywords: Dunaliella salina, extreme halophiles, habitability of Mars, hypersaline

brines, limits-of-life on Earth, solar salterns

**Topic:** biomass, ecosystems, sodium chloride

**Issue Section:** Review Article

#### INTRODUCTION

NaCl-saturated environments occur throughout Earth's biosphere, spanning a wide range of scales from solar salterns, inland salt lakes and deep-sea brines to fluid inclusions of NaCl crystals and liquids-of-deliquescence on halite. They also include ephemeral habitats such as those created by the sea spray which lands on rock surfaces or saline aerosols which circulate within the atmosphere. Marine solar salterns, which are of human construction and include NaCl-saturated crystalliser ponds, occur along the coasts of many countries with warm sunny climates, such as Australia, China, Indonesia, Japan, Spain and the USA (Antón *et al.*2000; Zhaxybayeva *et al.*2013; Cui and Qiu 2014; Mori *et al.*2016), while inland solar salterns are often situated where natural saline waters and high evaporation rates occur, such as Australia, India, Spain, Peru

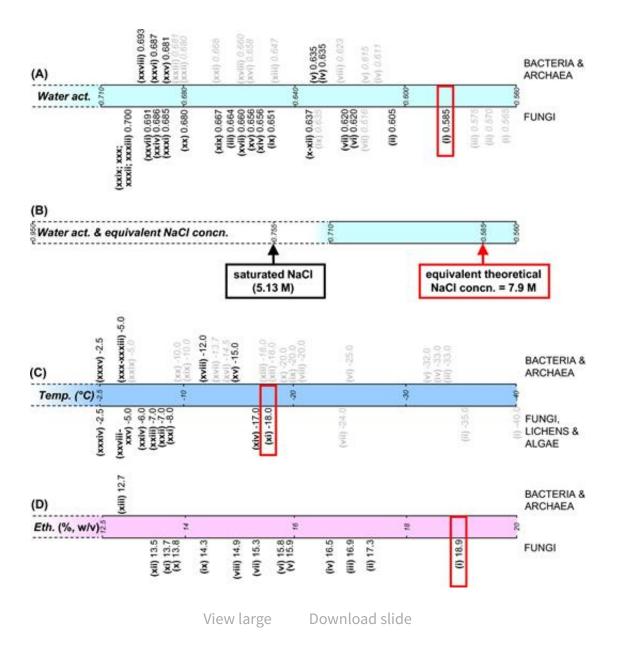
and the USA (Maturrano *et al.*2006; Zafrilla *et al.*2010; Jose and Jebakumar 2013). Large-scale inland bodies of water, including the Dead Sea (Rift Valley, Middle East) and the Great Salt Lake (Utah, USA), vary in their solute composition, but their NaCl component is at times saturated or oversaturated. In the Mediterranean Sea there is a cluster of deep-sea basins (e.g. L'Atalante and Bannock), most of which contain NaCl-saturated brines that are known amongst microbiologists as 'deep-sea hypersaline anoxic brine lakes'. NaCl crystals can form in these, as well as other, hypersaline systems. Ancient salt crystals make up evaporite sedimentary rocks that underlie some of Africa, Asia, Australia, Europe, and North and South America (Warren 2016); halite is also present at the surface and has been well studied, for example, in the Atacama Desert in Chile (Crits-Christoph *et al.*2016) and Death Valley in California (Lowenstein and Hardie 1985).

Many of these systems are geochemically complex, containing other types of ions such as  ${\rm Mg}^{2^+}$  and  ${\rm SO_4}^{2^-}$  (e.g. Yakimov *et al.*2015). Shallow saline lakes in particular are characterised by fluctuating salinity due to cycles of flooding, evapoconcentration and desiccation (Lowenstein and Hardie 1985). For instance, the shallow acid-brine lakes of Western Australia can cycle between ~5% and 35% w/v total salts and so are composed of either saline waters or hypersaline brines (Benison et al.2007). Natural brines tend to be enriched in many elements besides Na and Cl; these most commonly include S, Mg, Ca, K and Br, though brines with unusually high or low pH also may have elevated dissolved concentrations of Al, Fe and Si (Bowen and Benison 2009; whereas such elements are often present at trace levels, they can also reach concentrations which are stressful and/or toxic in some brines (see below). All brines are characterised by low water activity, a parameter that can induce potent stresses in microbial cells. Indeed, water activity (i.e. the effective mole fraction of water) acts as a determinant for life processes and habitability of hostile environments (Grant 2004; Rummel et al.2014; Stevenson *et al.***2015a**,**b**). This thermodynamic parameter, defined by Raoult's Law, is dependent on temperature and pressure. At the levels of cellular stress biology, and microbial ecology, water activity also interacts with factors such as temperature, pH and chaotropicity to determine rates and windows for metabolic activity and growth of microbial systems (e.g. Williams and Hallsworth 2009; Harrison et al.2013; Alves et al.2015).

Here, we focus on NaCl-dominated brines that are at or close to saturation with respect to halite, either all or some of the time. These systems are commonly regarded as representing the dry extreme for life—i.e. a limit of the Earth's biosphere—or used as a

model for studies of water-constrained ecosystems and habitability of hostile terrestrial environments in the context of astrobiology (e.g. Brown 1990; Grant 2004; Harrison *et al.***2013**; Crits-Christoph *et al.***2016**). In addition, there has been an assumption that the water-activity limit for the cells and ecosystems of the most extremely halophilic microbes is 0.755, i.e. the value corresponding to water that is saturated with NaCl (Anderson 1954; Brown 1976; Grant 2004; Kminek et al.2010; Rummel et al. 2014). However, recent studies have revealed active cell division of halophilic *Archaea* and *Bacteria* in the range 0.748-0.635 water activity, with theoretically determined water-activity minima as low as 0.611 (Stevenson et al.2015b). One study, of Aspergillus penicillioides (a xerophilic ascomycete capable of growing at high concentrations of various solutes including NaCl, sugars and glycerol), established a new water-activity limit for microbial differentiation and cell division: 0.585, with a theoretically determined limit of 0.565 (Fig. 1A; Stevenson et al.2017a,b,c). Collectively, these studies suggest that some extreme bacterial and archaeal halophiles may grow even when water activity is in the range 0.611-0.565. Furthermore, it is plausible that microbial populations which subsist under extreme conditions may undergo cell divisions over long timescales (Rummel et al.2014), yet to be quantified, at even lower water-activity values. Whereas the most-active microbes present in NaClsaturated environments are known as halophilic extremophiles, the water activity of NaCl-saturated brine falls far short of the limit for extant halophilic life or the theoretical limit to which microbes could plausibly adapt (see below). This article reviews the evidence that microbially diverse, biomass-rich, and highly productive ecosystems occur at 5 M NaCl, and that they are functioning under moderate, rather than thermodynamically extreme, conditions.

Figure 1.



The most extreme values thus far reported for microbial cell division based on empirical data (black values); microbial cell metabolism based on empirical data (grey values, normal font); and microbial cell division based on theoretically determined limits (grey values, italics) in relation to: (**A** and **B**) low water-activity (turquoise scale-bars), (**C**) sub-zero temperatures (blue scale-bar), and (**D**) ethanol concentration (pink scale-bar). The scale bars span 0.710 to 0.560 water activity, 0.950 to 0.560 water activity, -2.5 to -40°C, and 12.5 to 20% w/v ethanol, respectively. Red boxes indicate the empirically determined limits for microbial cell division of the most extremely stress-tolerant microbial system in relation to each stress. For (**A**), the microbial systems are: (**i**) *Aspergillus penicillioides* JH06THJ (Stevenson *et al.*2017c), (**ii**) *Xeromyces bisporus* FRR 0025 (basionym *Monascus bisporus*) (Pitt and Christian 1968; extrapolated value from Stevenson *et al.*2017b), (**iii**) *Xerochrysium xerophilium* FRR 0530 (formerly *Chyrsosporium xerophilum*) (Stevenson *et al.*2017b), (**iv**) haloarchaeal strain GN-5 and (**v**) haloarchaeal strain GN-2 Javor 1984; Stevenson *et al.*2015a), (**vi**) *Aspergillus echinulatus* (Snow 1949), (**vii**) *Zygosaccharomyces rouxii* (von Schelhorn 1950; extrapolated value from Stevenson *et al.*2015a), (**viii**) *Halorhodospira halophila* DSM 244<sup>T</sup> (Deole *et al.*2013; Stevenson *et al.*2015a), (**ix**) *Eurotium halophilicum* FRR 2471, (**x**) *X. bisporus* FRR 3443, (**xi**) *X. bisporus* FRR 1522 (**xii**) *X. bisporus* FRR 2347 (**ix**-

**xii** from Stevenson et al.2017b), (**xiii**) Halorhabdus utahensis DSM 12940<sup>T</sup> (Wainø et al.2000; Stevenson et al.2015a), (xiv) Eurotium amstelodami FRR 2792 and FRR 0475 (Armolik and Dickson 1956), (xv) Eurotium chevalieri JH06THI (Ayerst 1969), (xvi) Halobacterium strain 004.1 (Stevenson et al.2015a), (**xvii**) X. xerophilium CBS 153.67<sup>T</sup> (Pitt et al.2013), (**xviii**) Actinopolyspora halophila ATCC 27976T (Yoshida et al.1991; Stevenson et al.2015a), (xix) Eurotium repens JH06JPD (Stevenson et al.2017b), (xx) A. penicillioides in complex substrates (Hocking 2003), (xxi) Halanaerobium lacusrosei DSM 10165 (originally published as Haloanaerobium lacusroseus) (Cayol et al. 1995; Stevenson et al. 2017b), (xxii) Halorhodospira halochloris (basonym Ectothiorhodospira halochloris) (Deole et al. 2013; Stevenson et al.2015a), (xxiii) mixed halophile community (Antón et al.2000; Stevenson et al.2015a), (xxiv) Bettsia fastidia FRR 77 (basionym Chrysosporium fastidium) (Pitt et al. 2013), (xxv) Natrinema pallidum NCIMB 777 (Antón *et al.*2000; Stevenson *et al.*2015a), (**xxvi**) *Halobacterium noricense* DSM 15987 (Stevenson et al.2015a), (xxvii) Wallemia sebi FRR 1473 (Pitt and Hocking 1977), (xxviii) Halcococcus salifodinae DSM 13046 (Stevenson et al.2015a), (xxix) Eurotium rubrum FRR 0326 (Gock et al.2003), (xxx) Polypaecilum pisce FRR 2733 (Wheeler, Hocking and Pitt 1988), (xxxi) Eurotium repens FRR 382 (Stevenson et al. 2017b), (xxxii) Aspergillus conicus (Pelhate 1968), (xxxiii) Eremascus albus (Pitt 1975). For (B), the limit of NaCl solubility is indicated (at 0.755 water activity) and the theoretical solubility limit of NaCl is indicated at the 0.585 water-activity limit for microbial growth; see (A). For (C), the microbial systems are: (i) Pleopsidium chlorophanum (de Vera et al.2014), (ii) a mixed culture of Geomyces and Leucosporidium (Panikov and Sizova 2006), (iii) Paenisporosarcina sp. B5 (Bakermans and Skidmore 2011), (iv) Chryseobacterium sp. V3519–10 (Bakermans and Skidmore 2011), (v) Nitrosomonas cryotolerans ATCC 49 181 (Miteva, Sowers and Brenchley 2007), (vi) Planococcus halocryophilus Or1 (Mykytczuk et al. 2013), (vii) Cladonia foliacea (basionym Cladonia alcicornis) (Lange 1965), (viii) Colwellia psychrerythraea 34H (Junge et al. 2006), (ix and x) bacterial communities obtained from permafrost (Tuorto et al.2014 and Rivkina et al.2000, respectively), (xi) Rhodotorula glutinis FMT157 (Collins and Buick 1989), (xii) a mixed culture of Geomyces and Leucosporidium (Panikov and Sizova 2007), (xiii) Psychrobacter frigidicola ACAM 304<sup>T</sup> (Bowman et al.1996), (xiv) Umbilicaria aprina Nyl (Schroeter et al.1994), (xv) Planococcus halocryophilus Or1 (Mykytczuk et al.2013), (xvi) Colwellia psychrerythraea 34H (Huston 2003), (xvii) genetically modified Escherichia coli (Ferrer et al. 2003), (xviii) Psychromonas ingrahamii GVU73721 (Breezee, Cady and Staley 2004), (xix) Psychrobacter urativorans ACAM 534 (Bowman et al. 1996), (xx) permafrost Archaea (Rivkina et al. 2007), (xxi) Nitzschia frigida (Aletsee and Jahnke 1992), (xxii) Mrakia frigida (Larkin and Stokes 1968), (xxiii) Mrakia nivalis (Larkin and Stokes 1968), (xxiv) Thalassosira antarctica (Aletsee and Jahnke 1992), (xxv) Chlamydomonas sp. (Eddie, Krembs and Neuer 2008), (xxvi) Thyphula ishikariensis, (xxvii) Thyphula incarnata, (xxviii) Thyphula phacorrhiza (xxvi-xxvii from Snider et al.2000), (xxix) Methanogenium frigidum (Franzmann et al.1997), (xxx) Glaciecola punicea DSM 14233 (Chin et al.2010), (xxxi) Psychrobacter glacincola DSM 12194<sup>T</sup> (Chin et al.2010), (**xxxii**) P. urativorans ACAM 534 (Chin et al.2010), (**xxxiii**) Sporosarcina psychrophila DSM 3<sup>T</sup> (Chin et al.2010), (xxxiv) Humicola marvinii (Larkin and Stokes 1968), and (xxxv) Methanococcoides burtonii DSM 6242 (Franzmann et al. 1992). For (**D**), the microbial systems are: (i) Saccharomyces cerevisiae NCYC 1327 (Thomas et al. 1993), (ii) several strains of S. cerevisiae (Gray 1941), (iii) S. cerevisiae shochu strain 1200 (Ohta, Hamada and Nakamura 1993) (iv) S. cerevisiae IFO 2347 (Hayashida and Ohta 1981), (v) S. cerevisiae baker's strain (Carlsen, Degn and Lloyd 1991), (vi) S.

cerevisiae John Labbatt Ltd. 67 (Odumeru et al.1992), (vii) *S. cerevisiae* PE-2 and CA1185 (Nakamura et al.1996), (viii) *S. cerevisiae* brewer's strain (Jones, Thomas and Ingledew 1995), (ix) *S. cerevisiae* brewer's strain (D'Amore et al.1990), (x) *S. cerevisiae* brewer's strain (Nagodawitbana and Steinkraus, 1976), (xi) *S. cerevisiae* BAW 6 (Omori et al.1996), (xii) *S. cerevisiae* ale strain (Stewart et al.1988), and (xiii) *Zymomonas mobilis* (Seo et al. 2005; Cray et al.2015)

# THERE IS NO MICROBIAL ECOLOGY NEAR THE BIOPHYSICAL LIMITS FOR LIFE

In this article, we use the term 'extreme' to refer to conditions which impose nearcomplete constraints on cellular life and, as such, represent a thermodynamic fringe for microbial life on Earth. For instance, the water-activity window for the microbial biosphere spans from 1 to 0.585, i.e. 0.415 units. Only five or six types of microbe are known to function in the range 0.635 to 0.585; this range represents the most extreme 12% of the known window for microbial functionality (Fig. 1; Stevenson et al. 2015a, b). Various parameters, either singly or in combination, can prevent microbial cell function; regardless of species. These include extremes of temperature and/or pressure, pH, chaotropicity, ionic strength and low water activity. The biophysical mechanisms by which this occurs operate at the levels of the biomacromolecule, whole cell, ecosystem and, ultimately, the microbial biosphere (Kashefi and Lovley 2003; Hallsworth et al.2007; Chin et al.2010; Cray et al.2015; Santos et al.2015; Yakimov et al.2015; Stevenson et al.2015a,b; Fox-Powell et al.2016). For such parameters, the failure points for cell division of the most resilient microbial systems occur close to, for example, 121°C, -18°C, 18.9% w/v (24% v/v) ethanol, and 0.585 water activity (Fig. 1). In relation to water activity, Pitt and Christian (1968) stated that germination of Xeromyces bisporus (FRR 0025) aleurospores occurred at 0.605, a finding which has neither been surpassed nor repeated for this species (Fig. 1A; Williams and Hallsworth 2009; Stevenson and Hallsworth 2014; Stevenson et al. 2015a, b, 2017a, b). At this value, biotic activity was somewhat slow; germination was recorded only after a period of 3 months. The study of A. penicillioides strain JHo6THJ, which germinated under some conditions at water activities of <0.605, revealed more rapid growth (Fig. 1A; Stevenson et al.2017c). Even 0.585 water activity, after 4 days conidia had swollen by >40% on a volume basis and were in a pre-germination phase; by 11.5 days conidia had differentiated to form polarised, tapered germination structures; and between 36 and 57 days fully formed separate germlings had been produced (Stevenson et al.2017c).

In relation to low temperatures, there are few reliable reports of microbial growth in the range -5 to -15°C (Fig. 1D; Chin et al.2010; Rummel et al.2014). Beyond this range, there is a single report of microbial reproduction, for Rhodotorula glutinis, at  $-18^{\circ}$ C, although it took 17–18 weeks for cell numbers to double, and the yeast has an optimum temperature for growth between 20°C and 30°C (Fig. 1D; Collins and Buick 1989; Martínez 2006). Ethanol, which is produced in bulk by Saccharomyces cerevisiae and other microbes and acts as a potent antimicrobial agent, reduces water activity and acts as a chaotropic stressor for cellular systems (Hallsworth, Nomura and Iwahara 1998; Hallsworth, Heim and Timmis 2003; Cray et al.2013a, 2015). There are very few studies reporting microbial tolerance to ethanol at concentrations of >12.5% w/v (15.8% v/v) (Fig. 1C). However, one study demonstrated metabolic activity of, and ethanol production by, S. cerevisiae at 18.9% w/v ethanol (Fig. 1C; Thomas et al.1993), which corresponds to a water activity of 0.901 and chaotropic activity of  $24.31 \, \text{kJ kg}^{-1}$ (Hallsworth and Nomura 1999; Cray et al. 2013b). Reduced temperatures and kosmotropic substances, such as the compatible solute proline, can mitigate against and, in some cases, prevent the chaotrope-induced stress associated with ethanol and mechanistically comparable stressors (Hallsworth 1998; Hallsworth et al. 2007; Bhaganna et al.2010; Bell et al.2013; Alves et al.2015; Cray et al.2015). Kosmotropic substances are those which entropically order or stabilise biomacromolecular systems (Collins 1997; Cray et al. 2013b; Ball and Hallsworth 2015). Recent studies were carried out based on genome analysis, hybridisation and subsequent strain selection to optimise ethanol tolerance and product yield from S. cerevisiae strains used for bioethanol fermentations (Fig. 1C; Swinnen et al.2012; Pais et al.2013). Whereas these were successful, the ethanol concentrations obtained are some way below the 18.9% w/v value reported previously. In relation to the limit for microbial life, each of these extremes will be further examined below.

At or close to such thermodynamic boundaries for life, there is no evidence of any functional microbial ecosystem. Furthermore, there are no reports of microbial metabolism below 0.585 water activity or above 18.9% w/v ethanol (Fig. 1A), and only sporadic reports of limited metabolic activities in a handful of microbial species below –18°C (Fig. 1D; Kminek *et al.*2010; Rummel *et al.*2014). For instance, studies of the lichen *Pleopsidium chlorophanum* that were performed in a Mars Simulation Chamber indicated significant photosynthetic activity at –40°C (Fig. 1D; de Vera *et al.*2014). In natural habitats such as sea ice and permafrost, however, there is no evidence of ecosystem function at temperatures below –18°C; hence, the standard (–20°C) temperature of freezers used for food storage. The data arising from studies of the

lower temperature limits for microbial life have been analyzed elsewhere (Chin et al.2010; Kminek et al.2010; Rummel et al.2014) so will not be discussed at length here. There is no evidence of microbial fermentation at ethanol concentrations >18.9% w/v from culture-based studies; similarly, no anthropogenic system or microbial habitat is known to facilitate ecosystem function above such ethanol concentrations according to either culture-dependent or independent studies (Fig. 1C; Cray et al.2015; Lievens et al.2015). The chaotropicity of MgCl<sub>2</sub> also prevents the biotic activity of (all) microbial systems at concentrations  $\geq$  2.40 M (Hallsworth *et al.*2007), though activity for some halophile systems has been reported in the range 2.50-3.03 M when kosmotropic ions are present (Hallsworth et al.2007; Yakimov et al.2015). Whereas extrapolations made using data of fungal germination at low water activity suggest theoretical wateractivity minima for germination in the range 0.575-0.565 (Fig. 1A; Stevenson et al.2017b,c), there is no empirical evidence for metabolic activity or cell division of xerophilic fungi in this range. The few microbes known to be capable of biotic activity at < 0.690 water activity inhabit diverse substrates and environments in nature and so are unlikely to occur together or form an active community under these conditions (Fig. 1A; Stevenson et al.2015a,b, 2017c). In other words, conditions or habitats of < 0.690 might be regarded as 'extreme' and are characterised by low biodiversity and a negligible biomass of active microbes. Each of the findings which demonstrated microbial activity at -18°C, 18.9% w/v ethanol or 0.585 water activity was reported in an individual published study and for a single microbial system (Fig. 1; Collins and Buick 1989; Thomas et al.1993; Stevenson et al.2017c), and there is no functional ecosystem in any of these contexts. Indeed, chaotropic environments containing high concentrations of MgCl<sub>2</sub> or ethanol are functionally sterile (hence the value of such substances as food preservatives and/or biocides) (Hallsworth et al.2007; Cray et al.2015; Yakimov et al.2015).

## RICH AND ACTIVE ECOSYSTEMS IN NaCl-SATURATED BRINES

NaCl-saturated systems are highly permissive for microbial life (Table 1). This is also evidenced by the active international community of scientists working on these ecosystems; and the recent and well-attended 13th International Conference on Halophilic Microorganisms of the International Society for Salt Lake Research (2017, in Ulan-Ude, Russia with 122 delegates). There are 2569 papers published on halophiles

and their hypersaline habitats, according to the Thomson Reuters Web-of-Science database; a total that was derived (28 May 2018) using the following searches: '(halophil\*\* or halotoler\*\* or microbial community or microbial ecology or microbiology or *Haloquadratum* or *Dunaliella* or *Salinibacter* or *Actinopolyspora* or *Halorhodospira*) and (saltern\*\* or NaCl-crystal\*\* or halite or brine-lake\*\* or deep-sea-lake\*\* or deep-sea-basin\*\* or deep-sea-brine\*\* or fluid-inclus\*\* or crystallizer-pond\*\* or crystalliser-pond\*\* or Dead-Sea or hypersalin\*\* or NaCl-saturat\*\* or salt-saturat\*\* or saturated-NaCl or saturated-salt\*\* or 5-M-NaCl or 5-molar-NaCl)' and 'extreme-halophil\*\* or obligate-halophil\*\*.

**Table 1.**Microbiology of diverse, NaCl-saturated habitats.

Habitat	Microbial biomass	Phylogenetic diversity	Typical members of the community
Crystalliser ponds of marine solar salterns	Compared with most other aquatic microbial habitats, NaCl-saturated solar salterns are biomass rich. Even at ~5 M NaCl, the high cell density of halophilic microbes, and the high intracellular concentrations of their pigments, result in highly coloured turbid brines (Oren 2009). Cell densities are typically in the range 10 to 10 cells ml (Guixa-Boixareu et al.1996; Çınar and Mutlu 2016), which is orders of magnitude higher than those observed in a study of coral atolls in the Northern Line Islands (Dinsdale et al.2008). Indeed, the cell densities of hypersaline crystalliser ponds are comparable to, or even surpass, those of anthropogenic systems such as aquaculture waters, fishponds and ethanol-producing fermentations (Cocolin, Bisson and Mills 2000; Herrero et al.2006; Kalcheva	Marine, NaCl-saturated solar salterns are characterised by an exceptionally high microbial diversity, and contain taxa from each domain of life as well as diverse viruses. One study, which used both culture-based and culture-independent methods (Burns et al. 2004), revealed that an individual, NaCl-saturated saltern (coast of Australia) contained 57–66 identifiable types of Archaea and Bacteria. A 16S rRNA gene clone-library study of a Tunisian saltern (S5 pond, 31% total salinity), carried out by Baati	The archaeal genera that are commonly present include Haloarcula, Halobacterium, Halobaculum, Halococcus, Haloferax, Halobacterium, Halorubrum and Haloquadratum, and bacterial genera include Halomonas, Halovibrio, Salicola and Salinibacter spp. (Manikandan, Pašić and Kannan 2009; Sabet et al.2009; Cray et al.2013a). Eukaryotes include fungal genera such as Aspergillus, Cladosporium, Hortaea, Penicillium and Wallemia spp. (Cantrell, Casillas-Martínez and Molina 2006). The most common algae are Dunaliella spp., which are also the most xerophilic of the microalgae (Cifuentes et al.2001; Oren 2014;

*et al.*2008; Taniasuri, Lee and Liu 2016; Wada *et al.*2016).

et al. (2008), revealed eight OTUs pertaining to Bacteria (Bacteroidetes and Proteobacteria) and 39 OTUs assigned to Euryarchaeota (Halobacteria). For prokaryotes from this saltern, the Shannon Index (Spellerberg and Fedor 2003) was 0.98 for Bacteria and 3.04 for Archaea (Baati et al.2008). Another study retrieved 86 isolates of 26 fungal species from the waters of the Cabo Rojo salterns on the coast of Puerto Rico (Cantrell, Casillas-Martínez and Molina 2006); and a study of a NaCl-saturated crystalliser pond (Spanish coast) revealed the presence of 42 viral genomes (using a fosmid metagenome library) (Garcia-

Stevenson et al. 2015b). The nanoflagellates Euplaesiobystra, Halocafeteria, and Pleurostomum spp. are the most xerophilic protists known, and have also been found inhabiting NaCl-saturated salterns (Park et al. 2009; Stevenson et al.2015b). Amongst the microbes present in marine solar salterns, there are several species which recurrently dominate halophile communities, most notably Dunaliella salina, Salinibacter ruber and Haloquadratum walsbyi (Cray et al.2013a; Oren and Hallsworth 2014).

Liquids-ofdeliquescence on halite rock<sup>a</sup> Halite rock is commonly found in terrestrial surface and subsurface environments, and, under specific conditions, deliquesces to produce hypersaline NaCl brines (Rummel et al.2014). These brines, typically exposed to the air, are commonly inhabited by microbes. Cell densities reported for liquids-of-deliquescence on halite rocks of

The phylogenetically diverse communities present include microbes from each domain of life, from photosynthetic bacteria and algae to heterotrophic archaea and bacteria, including both aerobes and the anaerobic

Heredia et al.2012).

Microbial taxa found in liquids-of-deliquescence include all domains of life but archaeal species are typically the most prevalent (Robinson et al.2015). Of the DNA extracted and analysed, 71% of metagenomic reads were assigned to *Archaea* (the majority of which were Halobacteria),

the Atacama Desert are in the range  $2-10 \times 10^7$  cells  $g^{-1}$ , a microbial biomass which is three orders of magnitude greater than that of the surrounding soil (Robinson *et al.*2015).

fermentative bacteria of Halanaerobiales (Robinson et al.2015; Crits-Christoph et al.2016). The microbial diversity of liquids-ofdeliquescence can be extraordinarily high, the brine communities on halite of the Atacama Desert have Shannon Indices as high as 6.25, which is in excess of those usually associated with soils (Fierer and Jackson 2006). At least 30 distinct viruses were identified within the cellular biomass found at one sample site; most of these are thought to occur within cells of the Halobacteria (Crits-Christoph

27% to *Bacteria* and 1% to eukaryotes (Crits-Christoph et al. 2016). The most prevalent Archaea were, in descending order, Natronomonas, Halococcus, Haloarcula, Halorhabdus, Salinarchaeum, Halobacterium and Halomicrobium, and the most prevalent bacteria were Salinibacter. Cyanobacteria of the genera Chroococcidiopsis and Halothece carry out the majority of primary production in halite ecosystems (Wierzchos, Ascaso and McKay 2006; Robinson et al. 2015).

Deep hypersaline anoxic basins NaCl-saturated, deep-sea brine lakes are characterised by a considerable, highly active biomass (Daffonchio et al. 2006). The halocline at the interface of the deep-sea Lake Bannock (located beneath the Mediterranean Sea) and the open ocean has a NaCl concentration ranging from that of the overlying sea water (690 mM), where there are  $4.32 \times 10^4$ cells ml<sup>-1</sup>, to ~4.5 M which has  $4.39 \times 10^5$  cells ml<sup>-1</sup> (Daffonchio et al.2006). There is, therefore, an order-of-magnitude increase

Microbial diversity is high in deep-sea basins: the major eukaryotic groups represented include the opisthokonts, plants, cercozoa, alveolates, stramenopiles and discicristates (Edgcomb et al.2009); cultureindependent studies of Lakes Bannock and Thetis have identified from 40 to

et al.2016).

Insights into composition (and function) of microbial assemblages inhabiting the deep-sea brine lakes have been gained mostly by metagenomic analyses. The most-common Archaea in the brine layer of the Mediterranean deep-sea hypersaline lakes are MSBL1 and Halophilic Cluster 1 (HC1) candidate divisions (Daffonchio et al. 2006; La Cono et al.2011; Yakimov

in microbial biomass as NaCl concentration increases, a phenomenon reported within other deep-sea basin systems, such as the NaCl-saturated Lake L'Atalante (Mediterranean Sea) where a cell number of 1.07 × 10<sup>5</sup> ml<sup>-1</sup> has been recorded (i.e. more than that for the overlying sea water, which has 3.57 × 10<sup>4</sup> ml<sup>-1</sup>) (Yakimov *et al.*2007).

120 OTUs for protists alone (Edgcomb et al.2009; Stock et al.2012). Analysis of a Lake Thetis microbial community revealed that 37% of the total number of the OTUs were fungi (Stock et al.2012). Additional studies have revealed 71 unique sequences corresponding to Archaea and Bacteria (in a of more or less 1:1) in Lake Thetis (Ferrer et al. 2012). Methane oxidisers, methylotrophs, sulphate reducers, sulphur oxidisers and/or microaerophiles are found in Lakes L'Atlante, Bannock and Thetis (Daffonchio et al.2006; Yakimov et al.2007; La Cono et al.2011).

et al.2013). The vast majority of the bacteria inhabiting deep-sea brine lakes was assigned to Alpha-, Delta-, Epsilonand Gammaproteobacteria and KB1 candidate group (Yakimov et al.2007; La Cono *et al.*2011; 2013; Pachiadaki et al. 2014). Bacteria from the Desulphobulbaceae and Sphingobacteria have been identified in Lake Bannock (Daffonchio et al.2006). The mostcommon protistan genera in Lake Thetis include Pleuronema. Strombidium and *Trimyema*, while the predominant fungal RNA gene sequences are closely related to those of the genera Aspergillus, Cladosporium, Malassezia, and Rhodotorula (Stock et al.2012; Bernhard et al.2014).

Inland salt lakes

Of the large-scale water bodies located inland, it is estimated that 45% are saline (Wetzel 2001), a value which is likely to rise due to global warming (Williams 2002). Lake Kasin, located in Russia, is a shallow, NaCl-saturated water body containing sufficient iron to support specialist, ironmetabolising microbes. Fe(II) ions are weak electron donors, compared with organic substrates for instance, and can limit the efficiency with which

A number of alkaline, NaCl-saturated lakes are located in the Wadi An Natrun Valley in Egypt, including Lakes Rizunia, Gaar, and Zugm, and are characterised by highly diverse communities made up of autotrophic and/or heterotrophic taxa belonging to each domain of life

Both Lakes Chaka and Kasin are dominated by *Archaea* belonging to Halobacteria. Whereas the genera *Haloarcula* and *Halosimplex* prevailed in Lake Chaka, *Halobaculum* spp. seemed to be abundant in Lake Kasin (Jiang *et al.*2006; Emmerich *et al.*2012). The dominant bacterial classes in Lake Kasin were *Bacilli* and *Clostridia*, with *Halothiobacillus* as the

cell-available energy is generated. Living at 5 M NaCl imposes a high energetic cost to the cell. Nevertheless, the NaClsaturated sediment of Lake Kasin contains  $\geq 1 \times 10^4$  ironmetabolising microbial cells g<sup>-1</sup> (and  $1.1 \times 10^6$  to  $6.7 \times 10^7$  total microbial cells g<sup>-1</sup>) (Emmerich et al.2012). Lake Chaka, which is located on the Tibetan Plateau at 3200 m above sea level, is 2-3 cm deep and NaCl saturated. Despite the high altitude and relative isolation from other NaCl-rich habitats, the Lake Chaka brine contains  $4.8 \times 10^6$ cells ml<sup>-1</sup> and the lake sediment contains  $4.2 \times 10^7$  cells ml<sup>-1</sup> (Jiang et al. 2006). The NaClsaturated brine of Aran-Bidgol Lake (Iran) has been shown to contain  $3-4 \times 10^7$  cells ml<sup>-1</sup> (Makhdoumi-Kakhki et al. 2012). Numerous salt lakes with appreciable depths (> 10 m) were reported in Central Romania (Baricz et al. 2014, 2015; Andrei et al. 2015, 2017). These permanently stratified bodies of water feature steady physicochemical gradients resulting in sharp niche partitioning and distinctive microbial communities, respectively. As a general characteristic, a less saline (6-20% total salinity) surface layer overlies a NaCl-saturated bottom stratum in such lakes. Estimation of prokaryotic cell densities by qPCR and DAPI indicated values ranging between 10<sup>5</sup> and 10<sup>6</sup> cells ml<sup>-1</sup> for *Bacteria* and  $10^7 - 10^8$  cells ml <sup>-1</sup> for *Archaea*, with the densest microbial population recorded

(Imhoff et al. 1979; Selim *et al.***2012**). A metagenomic study on Lakes Fazda, UmRisha and Hamra, which are all located in the Wadi An Natrun, reported 345 bacterial and 198 archaeal OTUs present in total, and distinct prokaryote communities within, and between, the lakes (Mesbah, Abou-El-Ela and Wiegel 2007). High microbial diversity is also prevalent in Lake Aran-Bidgol, which has been found to contain 37 OTUs corresponding to Archaea and Bacteria; 63% of these were unrelated to any taxa thus far identified (Makhdoumi-Kakhki et al.2012). Andrei et al. (2015) reported an increasing prokaryotic diversity along the salinity gradient in two meromictic salt lakes of Central Romania. The highest OTU values were retrieved from the NaCl-saturated layers of Fara Fund (144 archaeal and 330 bacterial OTUs) and Ursu lakes (116 archaeal and 459 bacterial OTUs). Phototrophs

most-common genus (Jiang et al. 2006). Common bacteria in Lake Chaka include Salinibacter ruber, and those within the genera Bacillus and Halanaerobium (Emmerich et al.2012). In the Wadi An Natrun Valley, members of the bacterial genus Halorhodospira dominated six hypersaline alkaline lakes, and other prevalent microbes within these lakes included the archaeal genus Halobacterium, the proteobacterium Chromatium, and the cyanobacterial genera Spirulina and Synechcooccus (Imhoff, Hashwa and Trüper 1978). This has been corroborated by a metagenomic study which also detected members of the bacterial genera Rhodobaca and Roseinatronobacter and the cyanobacterial genus Euhalothece (Mesbah, Abou-El-Ela and Wiegel 2007). The study on permanently stratified salt lakes in Romania pointed to clear nichebased differentiation of microbial communities inhabiting the water column. In the aphotic, anoxic and hypersaline zone of investigated lakes (~35% total salinity), Bacteria dominated the microbial communities, being represented by

in the most-saline lakes (Andrei et al.2015, 2017). The concentration and composition of salts within large-scale inland bodies of water including the Dead Sea (Middle East) and the Great Salt Lake (Utah, USA) can vary. The Dead Sea contains relatively high levels of chaotropic MgCl<sub>2</sub> (Hallsworth et al.2007), but is nevertheless characterised by high levels of microbial biomass during periodic blooming events (Oren 1985; Oren and Gurevich 1995). Biomass-rich microbial communities are also active in the Great Salt Lake, even when NaCl concentrations approach saturation (Post 1980).

(cyanobacteria and Dunaliella sp.) were also found in these stratified lakes (Máthé et al.2014; Andrei et al.2015). One hundred genera of Archaea and Bacteria were isolated from the NaCl-saturated northern arm of the Great Salt Lake (Weimer et al. 2009). An analysis of 16S rRNA gene clone libraries recovered from brine samples taken from Lake Kasin identified 290 sequences for bacteria and 231 sequences for Euryarchaeota and other Archaea (Emmerich

anaerobes classified within the genera Desulphovermiculus, Acetohalobium and *Halanaerobium*. In such conditions, *Archaea* were less abundant and mostly represented by Halobacteria (Halorhabdus sp. and Halorubrum sp.). It should be noted that substantial number of reads retrieved in this study related to candidate OP1 and OD1 bacterial clades (Andrei et al.2015). The most common eukaryotic species in many inland salt lakes is Dunaliella salina, among other Dunaliella species (Imhoff, Hashwa and Trüper 1978; Weimer et al. 2009).

Fluid inclusions within individual NaCl-evaporite crystals

Each year, many 100 millions tons NaCl crystals precipitate from evaporating brines located in both natural halophile habitats and industrial salterns. Halophiles are readily entrapped during crystal precipitation and remain in NaCl-saturated brine inclusions within the crystal (Norton and Grant 1988). The concentrations of cells in these fluid inclusions can be equal to or greater than those found in the evaporate brines from which they came, and can exceed 4 × 10<sup>8</sup> cells g<sup>-1</sup> (Norton and Grant 1988).

A cultureindependent study by Baati *et al.* (2010) identified 21 bacterial and 44 archaeal OTUs in a sample taken from the fluid inclusions of NaCl crystals from the benthic surface of a solar saltern in Sfax, Tunisia (on the Mediterranean coast). Microbes from each domain of life have been found within hypersaline inclusions of NaCl crystals, including Dunaliella spp.,

et al.2012).

It may be that any of the microbes which inhabit supersaturated saline systems could potentially become entrapped within NaCl crystals. Amongst the taxa thus far identified within fluid inclusions of NaCl crystals are Dunaliella salina, Archaea such as Haloarcula, Halobacterium, Haloferax, Halonotius, Halorubrum, Haloguadratum and *Natronococcus*, and the bacteria Salinibacter and Halobacillus (Norton and Grant 1988; Mormile et al.2003; Conner and Benison 2013; Henriet

Archaea and Bacteria et al.2014). (Lowenstein, Schubert and Timofeeff 2011; Conner and Benison 2013).

View Large

#### **Biomass density and richness**

In both natural and anthropogenic systems, including industrial salterns, NaClsaturated brines can have an extraordinarily high microbial biomass. NaCl-saturated solar salterns have 10<sup>6</sup> to 10<sup>8</sup> microbial cells ml<sup>-1</sup> (Table 1), and this is equal to the yeast-cell densities attained in wine and cider fermentations (Cocolin, Bisson and Mills 2000; Herrero et al. 2006; Taniasuri, Lee and Liu 2016). Even within the hypersaline fluid inclusions of a single NaCl crystal, cell densities greater than 4  $\times$  10  $^{8}$  ml  $^{^{-1}}$  have been observed (Table 1) where these communities can remain active; survive within the crystal over long time periods (possibly geological timescales); can be released upon dissolution of crystals or crystal deposits; and collectively constitute enormous microbial ecosystems which extend across considerable regions of the Earth's surface and subsurface (Grant 2004; Schubert et al.2009; Sankaranarayanan et al.2014). Furthermore, NaCl-saturated systems frequently exhibit a high abundance and diversity of microbial types (Table 1). The Shannon Indices in one marine saltern in Tunisia, for instance, were 0.98 for Bacteria and 3.04 for Archaea (Table 1; Baati et al.2008), values comparable to those for microbial diversity within bioethanol fermentations of corn, i.e. up to 3.23 (Li, Heist and Moe 2016). The salt-saturated bottom water layer (~35% total salinity) of two inland permanently stratified (meromictic) salt lakes situated in Romania revealed unusually high alpha-diversity for both *Bacteria* (SI 3.5 and 5.4; 330 and 459 operational taxonomic units—OTUs<sub>0.03</sub> – defined as ≥97% 16S rRNA gene sequence similarity clusters) and Archaea (SI 4.0 and 3.3; 144 and 116  $OTUs_{0.03}$ ). In these lakes, the microbial diversity increased along the salinity gradient reaching a maximum at the highest salt concentration (Andrei et al.2015). In Mediterranean deep-sea brines, despite the lack of phototrophs, more than 110 OTUs were present, encompassing microbes from each domain of life (the criterion ≥98% was used for 16S rRNA gene sequence similarity) (Table 1; Ferrer

<sup>&</sup>lt;sup>a</sup>In some environments, these habitats/ communities may occur simultaneously in the same location.

et al.2012; Stock et al.2012).

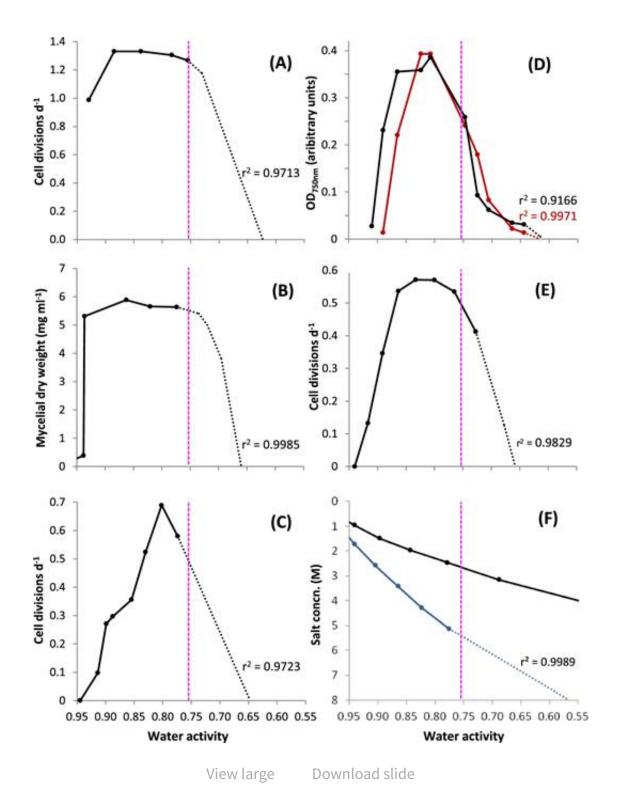
At specific relative humidities and temperatures, salts absorb water from the vapour phase and liquefy to produce saturated brine (Gough et~al.2011; Rummel et~al.2014; Stevenson et~al.2015b). The liquids-of-deliquescence on halite of the Atacama Desert can contain cell densities of  $2-10\times10^7$  cells  $g^{-1}$  (Table 1), and the NaCl-saturated north arm of the Great Salt Lake and saltern crystalliser ponds (in diverse locations) contain  $10^7-10^8$  microbial cells  $ml^{-1}$ , a density that gives rise to the opacity and distinctive red colour of these brines (Table 1; Oren 2009). Most other aquatic systems are characterised by microbial cell densities that are significantly lower than this. For example, a study of tropical freshwater reservoirs in Sri Lanka reported up to  $4.73\times10^6$  bacterial cells  $ml^{-1}$  (Peduzzi and Schiemer 2004) and a study of coral atolls reported from  $7.2\times10^4$  to  $8.4\times10^5$  microbial cells  $ml^{-1}$  seawater, depending on the sampling site (Dinsdale et~al.2008). A number of studies show that cell number increases as NaCl concentration increases (e.g. Benlloch et~al.1996; Daffonchio et~al.2006). There are various reasons which, when combined, help to explain why cell densities are so high in hypersaline environments; these include:

- I. salterns, and other NaCl-saturated habitats, can have a plentiful supply of nutrients (including the glycerol leaked by algae; Elevi Bardavid and Oren 2008; Elevi Bardavid, Khristo and Oren 2008; Oren 2017)) that can become further concentrated by evaporative processes,
- II. saturated NaCl is not thermodynamically extreme, and daytime temperatures (as well as water activity) are often close to the optimum for haloarchaea and other halophiles (see below, Oren 1994),
- II. the most-extreme halophiles are able to fully adapt to hypersaline conditions (Cray et al.2013a; Oren and Hallsworth 2014), and so their high growth rates and biomass are a reflection of the high nutrient levels,
- IV. salt concentrations and water activity typically remain more stable than those of saline waters that are not saturated, and
- V. the losses due to predation/grazing are relatively low (Guixa-Boixareu *et al.*1996; Pedrós-Alió *et al.*2000; Pedrós-Alió, Calderón-Paz and Gasol 2000).

### Halophile function at extremely low water activity

The water activity of NaCl-saturated brines (at ca. 0.755) is hostile to many types of microbe. Hundreds of known microbial species are likely capable of growth in the water-activity range 0.755-0.701, but below this range there is a more restricted number of microbial systems able to function (Fig. 1A; Williams and Hallsworth 2009; Stevenson *et al.*2015a,b, 2017c). Even so, some extremely halophilic prokaryotes which can be active down to 0.635-0.611—exhibit high or optimum rates of growth at saturated NaCl (Figs 1A and B and 2; Stevenson et al.2015a, b). For instance, the extremely halophilic archaeal strains GN-2 and GN-5 (both presumed to be Halobacterium), Halobacterium strain 004.1, Haloquadratum walsbyi (DSM 16790) and Halorhabdus utahensis (DSM  $12940^{T}$ ), and the bacteria Actinopolyspora halophila (ATCC 27976<sup>T</sup>), Halorhodospira halophila (DSM 244<sup>T</sup>), Pontibacillus strain AS2 and Salinicola strain LC26 exhibited very high or optimum growth rates at saturated NaCl or at wateractivity values equivalent to that of saturated NaCl (Fig. 2; Stevenson et al.2015b). Furthermore, their growth windows extend down to ~0.600 water activity, so more than one third of the water-activity window for growth lies beyond the water-activity value for saturated NaCl (Figs 1A, B and 2A-E). Similarly, for A. penicillioides, growth rates of germlings are remarkably high at ~0.755 water activity on media supplemented with NaCl and glycerol, and 42% of the water-activity window lies beyond this value; i.e. at water activities lower than (Stevenson et al.2017a,b,c). This finding is pertinent to the ecophysiology of halophiles because A. penicillioides, along with extreme halophilic fungi such as Hortaea and Wallemia species and halophilic algae such as Dunaliella species (all capable of growth at or close to 5 M NaCl, see Stevenson et al.2015b), are exposed to molar concentrations of glycerol which they accumulate in the cytosol at high salt concentrations.

#### Figure 2.



In relation to the water activity of saturated NaCl, i.e. 0.755 (pink dashed line), (**A–E**) plots are shown of doubling times, dry mass, or optical density measurements—i.e. turbidity—for extreme halophiles and (**F**) salt concentrations for NaCl (blue line) and MgCl<sub>2</sub> (black line; Hallsworth *et al.*2007 at 25°C). Growth curves for (**A**) the bacterium *Halorhodospira halophila* (strain DSM 244<sup>T</sup>, cultured in a defined medium supplemented with NaCl) (basonym *Ectothiorhodospira halophila*), (**B**) the bacterium *Actinopolyspora halophila* (strain ATCC 27976<sup>T</sup>, cultured in a complex medium supplemented with NaCl at 37°C), (**C**) the archaeon *Halorhabdus utahensis* (strain DSM 12940<sup>T</sup> in a defined medium supplemented with NaCl at 30°C), (**D**) haloarchaeal strains GN-2 and GN-5 shown in red and black, respectively (cultured in bittern

brines supplemented with peptone at 37°C) and (**E**) the archaeon *Halobacterium* strain 004.1 (in a synthetic seawater medium supplemented with NaCl, MgCl<sub>2</sub>, Na<sub>2</sub>SO<sub>4</sub> and KCl) were reproduced from Stevenson *et al.* (2015a); original data were obtained from Javor (1984), Yoshida *et al.* (1991), Wainø, Tindall and Ingvorsen (2000) and Deole *et al.* (2013). Extrapolations of growth curves (dotted lines) were carried out using Excel or Sigmaplot as described by Stevenson *et al.* (2015a). Plots of water activity versus salts concentrations (**F**) were modified from Hallsworth *et al.* (2007). The water activity versus concentration curve for NaCl was extrapolated using polynomial order 2 in Excel (blue dotted line).

Some experimental settings, as well as studies of bittern brines, represent a transition between Mg-rich environments that have no microbial activity, and densely populated NaCl-saturated brine (Javor 1984; Baati *et al.*2011; Yakimov *et al.*2015; Stevenson *et al.*2015b). Biotic activity of halophiles was thereby revealed in culture at <0.755. However, the environmentally relevant salts which reduce water activity to this level are typically chaotropic (see above), and so chaotropicity cuts in as the stress parameter to limit life before water activity can do so (Hallsworth *et al.*2007; Alves *et al.*2015; Stevenson *et al.*2015a,b; Yakimov *et al.*2015). Other substances, such as glycerol, are more permissive such that NaCl+glycerol mixtures enable microbial activity at <0.600 water activity (see above).

### **Ecosystem activity and complexity**

The microbial ecology of biomass-dense, NaCl-saturated habitats is both complex and dynamic (Table 1). Rates of primary production (via photosynthesis) of 120-220 µg carbon l<sup>-1</sup> d<sup>-1</sup> have been reported in crystalliser-pond brines (Elevi Bardavid, Khristo and Oren 2008), and the alga Dunaliella salina produces (and releases) sufficient amounts of glycerol to be able to sustain communities of heterotrophs (Hart and Gilmour 1991; Elevi Bardavid and Oren 2008; Elevi Bardavid, Khristo and Oren 2008; Oren 2017; Williams et al. 2017). Heterotrophic microbes are abundant in diverse hypersaline systems, including the haloarchaeal genera Halobacterium and Halococcus (Table 1), and the bacterium Salinibacter ruber which is commonly found in some NaClsaturated habitats (Antón et al. 2008; Cray et al. 2013a). Salinibacter ruber can utilise glycerol, converting some to dihydroxyacetone (glycerone) which is excreted, and may be utilised by halophilic Archaea such as Hqr. walsbyi (Sher et al. 2004; Elevi Bardavid and Oren 2008; Elevi Bardavid, Khristo and Oren 2008). Halophiles are known to be highly proficient at the symbiotic utilisation of various metabolites excreted by the primary producers, and the necromass produced by microbial cell turnover. This is exemplified both by the utilisation of glycerol leaked from microalgae and some deepsea NaCl-saturated systems, where communities can be sustained by trophic networks which ultimately rely on the organic matter that migrates from the overlaying seawater (Daffonchio *et al.*2006; Yakimov *et al.*2013). A recent review of the various roles of glycerol in the context of the ecological complexity of hypersaline habitats has been carried out by Oren (2017).

The various halophiles found in hypersaline brines exhibit a wide range of metabolic capabilities. For instance, inland hypersaline alkaline lakes located in Egypt contain microalgae, cyanobacteria, anoxygenic phototrophic bacteria, denitrifying bacteria and (in sediments at least) sulphate reducers (Table 1; Imhoff, Hashwa and Trüper 1978; Oren 2011). Microalgae (such as algae of the *Mamiellaceae* family) and cyanobacteria (*Halothece* spp.), as well as halophilic bacteria and archaea (e.g. *Halorhabdus* spp. and *Halococcus* spp.), inhabit the liquids-of-deliquescence according to studies of halite rock in the Atacama Desert (Robinson *et al.*2015; Crits-Christoph *et al.*2016). Another study of halite rock demonstrated carbon fixation in the absence of light, with ammonia used as an electron donor, further exemplifying the diversity of microbial activities which take place in the communities inhabiting liquids-of-deliquescence (Davila *et al.*2015).

Sulphate-reducing bacteria, such as Desulphobacter, Desulphovi-brio and Desulphococcus, can be active at NaCl saturation in the clay-, silt- and sand sediments of salterns (Kerkar and Loka Bharathi 2007; Mani et al. 2012); sulphur-oxidising chemolithotrophs of the family Ectothiorhodospiraceae are abundant in environments where NaCl concentration ranges from 1.7 to 4.3 M (León et al. 2014). Together, these findings indicate active bacterial networks for the redox cycling of various sulphur species. Anaerobic iron metabolism has been demonstrated in Lake Kasin (Russia) and Lake Chaka (China) (Jiang et al. 2006; Emmerich et al. 2012), while archaeal methanogenesis coupled with active sulphur cycling (by epsilonproteobacterial sulphur oxidisers and deltaproteobacterial sulphate reducers) has been reported to sustain biomass-rich complex microbial communities in deep-sea brine lakes such as those in the Urania basin of the Mediterranean (Borin et al. 2009) (N.B., in microbiology, the convention has arisen to refer to the bodies of hypersaline water located in deep-sea basins' as brine lakes', whereas geologists use the term' stratified marine brines' when referring to a body of brine which is overlaid by seawater). Active networking between sulphur-oxidising chemolithoautotrophs similar to the endosymbiont of the deep-sea geothermal vent tube worm Riftia pachyptila, and sulphate reducers, methanotrophs and nitrogen fixers, has been indicated by

metagenomic and metatranscriptomic analyses across the chemoclines of the deep Mediterranean Sea hypersaline anoxic lake Thetis (Ferrer *et al.*2012; Pachiadaki *et al.*2014). Other studies have focused on characterising the geochemical and biological complexity of microbial ecosystems, including biogeochemical cycling in NaCl-saturated deep-sea brines (Table 1; Daffonchio *et al.*2006; Alexander *et al.*2009; Edgcomb *et al.*2009; Danovaro *et al.*2010; Ferrer *et al.*2012).

Many inland alkaline saline lakes result from evaporation of Na-Cl-CO<sub>3</sub>-HCO<sub>3</sub>-rich waters. These brines precipitate minerals such as halite, calcite, and Na-carbonates such as trona, and halite. Alkaline hypersaline (haloalkaline) waters such as those in Wadi An Natrun (Egypt), Lake Sambhar (Rajasthan, India) or Salton Sea (California, USA) have high NaCl contents but low alkalinity. In this way, their water chemistry differs from that of soda lakes found in arid or semi-arid zones such as North America (e.g. Mono, Big Soda and Soap lakes), Central Asia (e.g. in the SE Siberian steppe and Inner Mongolia) as well as in the East African Rift Valley (e.g. Magadi, Bogoria and Natron lakes) (Sorokin et al.2014). These soda lakes are characterised by elevated  $\mathrm{HCO_3}^{-}/\mathrm{CO_3}^{2^{-}}$  concentrations. The haloalkaline Lake Sambhar is a large, inland, NaCland bicarbonate-dominated system where salt concentrations fluctuate depending on season and local rainfall. Total salt content within can reach up to 30% (w/v), with Na<sup>+</sup> at 118.1 g  $l^{-1}$ ,  $Cl^{-}$  at 147.12 g  $l^{-1}$  and  $HCO_3^{-}$  41.5 g  $l^{-1}$ . The remaining ions include other metallic ions  $Ca^{2+}$  (0.002 g  $l^{-1}$ ),  $Mg^{2+}$  (0.018 g  $l^{-1}$ ),  $K^{+}$  (1.96 g  $l^{-1}$ ), Fe (0.076 g  $l^{-1}$ ) and Mn  $(0.0052~{\rm g~l^{-1}})$  and anions  ${\rm NO_3}^-$  (1.07 g  ${\rm l^{-1}})$  and  ${\rm CO_3}^{2^-}$  (33.12 g  ${\rm l^{-1}})$  (Pathak and Cherekar 2015). The pH varies depending on the salinity, typically ranging from 8.18 to 11 (Yadav, Sarin and Krishnaswami 2007; Pathak and Cherekar 2015). One study revealed 64 morphologically distinct isolates in 26% w/v NaCl samples taken from Lake Sambhar, including the extreme halophile Natronococcus that is known to grow up to 32% w/v NaCl (Pathak and Cherekar 2015). It is populated by haloalkaliphilic *Archaea* (Upasani and Desai 1990), *Actinobacteria* (Jose and Jebakumar 2013), several members of the phylum Firmicutes (Paul et al. 2015), Proteobacteria (Cherekar and Pathak 2015), a number of halophilic or halotolerant bacteria (Bachani et al.2016), as well as phytoplankton that exhibit a range of salt tolerances (Jakher, Bhargava and Sinha 1990).

Despite their multiple stress parameters (salinity; low water activity; pH >9.5), both alkaline hypersaline lakes and soda lakes are among the most productive ecosystems, up to  $7 \, \mathrm{g \, C \cdot m}^{-2} \cdot \mathrm{day}^{-1}$  (Melack and Kilham 1974). Furthermore, they are inhabited by

biodiverse communities of microorganisms which drive full carbon-, nitrogen-, phosphorus and sulphur C-, N-, P- and S cycles (Jones et al.1998; Sorokin et al.2014; Sorokin, Banciu and Muyzer 2015). Soda lakes have, therefore, become a rich source for novel microbes during bioprospecting studies. Some of the microbes isolated from haloalkaline brines are capable of energy generation, e.g. via elemental sulphur disproportionation, acetoclastic and hydrogenotrophic methanogenesis, synthrophic methanogenesis (Sorokin et al. 2015, 2016a, 2017; Sorokin and Chernyh 2016), and degradation of recalcitrant polymers such as cellulose, chitin and pectin (Zhilina et al.2005; Sorokin et al.2012a and 2012b, respectively) regardless of the cellular inhibition imposed by the *in-situ* stress parameters. This said, there has been little work done on the water activity of haloalkaline environments. Whereas the NaCl concentration of hypersaline alkaline lakes can reach up to 5 M NaCl (e.g., in Wadi An Natrun), it is only 0.5-1 M in soda lakes, which contain more carbonates (~2 M Na<sub>2</sub>CO<sub>3</sub> and NaHCO3 in total). At in situ concentrations, sodium carbonate behaves as a weak electrolyte, and organisms inhabiting such environments require less energy for osmotic protection than organisms in NaCl-dominated, pH-neutral environments (Sorokin, Banciu and Muyzer 2015; Sorokin et al.2016a). Furthermore, the osmotic pressures of synthetic analogues for waters from soda lakes differ from those of pHneutral saline lakes at equivalent Na<sup>+</sup> concentrations because carbonate/bicarbonate ions have a less effect than Cl ions (Banciu et al. 2004). Collectively, these data suggest that the water activities of soda lakes, and many haloalkaline lakes, are considerably higher than 0.755.

The microbial communities discussed above (see also Table 1) are sufficiently diverse in terms of phylogeny and ecophysiological functions to form sustainable ecosystems. Their primary producers can support heterotrophic, and other trophic, networks without requiring inputs from outside the NaCl-saturated milieu. Furthermore, culture-independent studies indicate the presence of genes that are collectively capable of completing the biogeochemical cycles that can enable the microbial ecosystem to function independently (Allen and Banfield 2005). However, the ecological interactions of halophilic microbes found in NaCl-saturated habitats are neither limited to other microorganisms nor to the immediate locality. For instance, *Dunaliella* and *Asteromonas* and other microbial biomass support high densities of *Artemia* (brine shrimp) in saltmarshes, salterns and brine lakes (Triantaphyllidis, Abatzopoulos and Sorgeloos 1998; Clegg and Trotman 2002) which, in turn, may sustain populations of aquatic insects, and birds such as flamingos, gulls and waders

(Ayadi *et al.*2004; Torrentera and Dodson 2004). Such avian species by nature have a wide home range and/or are migratory, so microbial halophiles can ultimately impact biological systems located far from their hypersaline habitat. This exemplifies how the ecosystems found within many types of NaCl-saturated environment are inherently highly fertile and yet are not self-contained; i.e. that they can provide inputs into other, remote and/or non-saline ecosystems.

### NaCl-saturated brines as a hotspot for life

A NaCl-saturated body of water can act as a hotspot for life within in an otherwise low-biomass, low-biodiversity environment. Recent studies of NaCl-dominated lakes in Western Australia, for instance, exemplify this well; these lakes are located in arid, nutrient-poor regions characterised by sandflats and sand dunes. Even those lakes with a low pH (that are hostile due to both extreme acidity and the presence of ions that can be toxic to life) are populated by microbes and are surrounded by plants (including two species of *Eucalyptus* and *Enchylaena tomentosa*, the ruby saltbush), with animal populations also in evidence (Benison 2006). Moreover, the pH-neutral lakes (including those at NaCl saturation) are extremely rich in microbial biomass and diversity, with some also containing dense populations of *Artemia* as well as *Coxiella* gastropods and other invertebrates, and are surrounded by an abundance of plant and animal life including insects and spiders, snakes, emus and other birds, and mammals such as dingoes and kangaroos (Benison 2006).

The NaCl-saturated brines of Death Valley also support a biodiverse ecosystem, including plants of the genus *Salicornia*, some of which can grow while their roots are in NaCl-saturated brine, and gastropods such as the Badwater snail, *Assimnea infirma*. Incredibly, a fish *Cyprinodon salinus* (the pupfish) can also survive at molar concentrations of NaCl (Schoenherr 1992). In addition, many mammal, reptile, bird and insect species live near and interact with the saline waters within the valley (Bryan and Tucker-Bryan 2009). In terms of microbiology and, in many cases flora and fauna, many other types of NaCl-saturated brines act as oases for life in regions which are generally barren (Table 1; Daffonchio *et al.*2006; Crits-Christoph *et al.*2016; Davila *et al.*2015; Robinson *et al.*2015; Çınar and Mutlu 2016).

### Species dominance within the open habitats created by NaClsaturated brines

In some NaCl-rich environments, there is a shift in microbial habitability—a selection for extreme halophiles—at ~3.4 M NaCl. This is evidenced, for instance, by shifts in community structure in solar salterns, as inferred from the relative percentages of the various phylum- or class-level taxa present (Ventosa et al. 2015, and references therein). In a study of a series of saltern ponds at Santa Pola (Spain), those with lower NaCl concentrations were populated by taxa including Euryarchaeota (~46% of the community), Alphaproteobacteria, Actinobacteria, Verrucomicrobia, Deltaproteobacteria and *Firmicutes* (Ventosa *et al.***2015**). Those with 3.6–5 M NaCl were populated primarily by members of the Euryarchaeota (80–90% of the community) and Bacteroidetes (9– 10%). How the two dominant phyla manage to sustain viable ecosystems (for instrance, via what syntrophic synergies) is a topic worthy of future exploration. Shifts in microbial community composition from moderate- (<3.5 M) to high-salt (>3.5 M) conditions might be governed by a combination of several factors including the availability of light and, perhaps, oxygen, that simultaneously select for dominance of aerobic or microaerophilic haloarchaea and Bacteroidetes in hypersaline Santa Pola salterns. Impacts relating to light availability may be due to the albedo effect that can influence how much solar radiation is retained by shallow waters, according the colour of the underlying sediment. If the water is on a white salt, for example, the light is readily reflected by the sediment. For waters that are underlain by darker materials, such as a black mud or dark microbial biomass, more light is absorbed and so is retained within the water body. Abundant chemical precipitation of minerals can also make water turbid. For example, if a salt lake is precipitating clay minerals or iron oxides (which happens in Western Australian acid brine lakes), then the water gets cloudy temporarily, and light would not penetrate the water column as well.

A reverse trend was noted along the salinity gradient in a meromictic hypersaline lake in Romania. *Euryarchaeota* dominated (~90% relative abundance) the communities dwelling in the photic, oxic upper water-layer (20–25% total salinity), whereas the aphotic, anoxic and hypersaline bottom brine (~35% total salinity) was the niche preferred by anaerobic halophilic bacteria (*Deltaproteobacteria*, *Clostridia*, *Spirochaetes*, *Bacteroidetes* and *Actinobacteria*) in an overall abundance of ~75%. Under such conditions, the haloarchaea became secondary (~20%), and were mostly represented by *Halorhabdus* spp. (Andrei *et al.*2015). The structuring of microbial diversity at a global scale, however, may be driven primarily by salinity, and also by habitat type (water versus sediment) and lifestyle (free-living versus host-dependent) (Lozupone and Knight 2007; Auguet, Barberan and Casamayor 2010), and hypersaline sediments are habitats that can exhibit a phylogenetically diversity that is amongst the highest on

Earth (Table 1; Daffonchio et al. 2006; Lozupone and Knight 2007).

Some types of NaCl-saturated environments, such as salterns, represent open habitats that are characterised by high levels of competition between different microbes; they are resource rich and are available for colonisation, and are habitable, by a large diversity of microbes (Table 1; Hallsworth 1998; Cray *et al.*2013a; Oren and Hallsworth 2014; Lievens *et al.*2015). For a more detailed definition of open habitats, see Cray *et al.* (2013a). In addition, hypersaline systems in which NaCl concentration varies (due to processes such as ingress of freshwater through rivers or streams, precipitation and solar evaporation) are also open habitats for microbes during their transitions between different salt concentrations.

Such transitions can promote competition (Řeháková et al.2009) and, under these conditions, some halophiles reach high cell densities and become dominant species within the community; these species are known as 'microbial weeds' (Cray et al.2013a; Oren and Hallsworth 2014). In solar salterns, weed species include D. salina, S. ruber and *Hqr. walsbyi* (Cray et al.2013a), and can even dominate in low-temperature hypersaline systems such as Deep Lake in the Antarctic (DeMaere et al.2013). In this environment, the mean temperature is  $\sim -20^{\circ}$ C and yet haloarchaea, such as Halohasta litchfieldiae strain tADL which comprises 44% of the population, are still able to outcompete and outgrow potential competitors (Williams et al. 2014). Paradoxically, the halophilic archaea isolated from this lake are all mesophiles, with optimum growth temperatures of around 30°C. Haloferax mediterranei also has many characteristics of a weed species and, whereas it does not dominate microbial communities in saltern brines (Oren and Hallsworth 2014), it can do so in high NaCl sediments (Youssef, Ashlock-Savage and Elshahed 2012). Microbial weed species, therefore, may simultaneously cause some reduction in species diversity and an increase overall cell density (due to their own abundance). Whereas microbial weeds can in this way act as 'mavericks', the ecology of NaCl-saturated systems typically function in both a self-regulating and sustainable manner.

# CELLULAR ADAPTATIONS THAT ENABLE HIGH RATES OF BIOTIC ACTIVITY AT 5 M \$NaCl\$

Strategies to overcome high concentrations of salts

Halophile systems are readily able to function at high concentrations of NaCl because they have many adaptations to enable salt tolerance, and the water activity remains permissive for life. These adaptations are detailed in reviews by Grant (2004), Balashov et al. (2005), Oren (2013), Cray et al. (2013a) and Lebre, Maayer and Cowan (2017). Adaptations of halophiles to high salinity in the extracellular milieu include several mechanisms to balance the osmotic stress induced by such conditions. Indeed, for some halophilic microbes, NaCl-saturated brines may not significantly stress the cellular system (Stevenson et al. 2015b; Hallsworth 2018). A number of strategies adopted by halophiles to balance osmotic stress caused by high salt concentrations have been investigated, which allow classifying halophiles into either being 'salt-in', including halophilic archaea of the class *Halobacteria* and a small number of halophilic bacteria, or 'salt-out' represented by halophilic methanogenic archaea, most halotolerant/philic bacteria and all halophilic eukaryotes. Both these strategies involve a number of molecular adaptations that make it possible to adjust to the osmotic conditions of saline environments. Halophiles adopting salt-in strategies balance the low water activity of the extracellular milieu by accumulating compensatory amounts of K<sup>+</sup> sometimes accompanied by minor amounts of Na<sup>+</sup> ions (Oren 2008). In contrast, the salt-out strategy employed by most halophilic bacteria involves mechanisms for excluding salt from the cellular environment or the synthesis of compatible solutes, such as glycine betaine (i.e. N-trimethylated glycine) for bacteria and glycerol for eukaryotes (Roberts 2005; Youssef et al., 2014). Arguably, many extreme halophiles accumulate ions rather than organic compatible solutes for this purpose in order to conserve energy (Oren 1999).

In hypersaline environments, most proteins from non-halophilic microbes are inhibited due to the low water activity and/or high ionic strength, and are typically also denatured and aggregate and/or precipitate (Baldwin 1984; Li, Zhou and Mu 2012; Alves *et al.*2015). However, the extracellular and cytosolic enzyme systems of some halophiles can work optimally at NaCl concentrations as high as 5 M or concentrations of K<sup>+</sup> that are osmotically equivalent, respectively. Intracellular proteins of salt-in species exhibit increased hydrophilicity and charge allowing them to work at very high salt concentrations. Several studies, including those that first sequenced *Halobacterium* sp. NRC-1 (Reistad 1970; Ng *et al.*2000) and those that resolved the structure of halophilic proteins (Dym, Mevarech and Sussman 1995; Kennedy *et al.*2001; DasSarma *et al.*2006), revealed that, for some halophiles, their proteins contain few basic amino acids (relative to their orthologs in non-halophilic proteins), but are enriched in the acidic amino acids, glutamic acid and aspartic acid. They also revealed that halophilic

proteins of 'salt-in' strategists (Bolhuis, Kwan and Thomas 2008; Capes, DasSarma and DasSarma 2012) have more negative charges on their surfaces than comparable proteins from non-halophiles. Structural studies have also revealed that increasing solvation for halophilic enzymes makes them capable of performing their activities at (or below) 0.755 water activity (Karan, Capes and DasSarma 2012).

Indeed, some enzyme systems remain functional close to a water–activity value of zero (from microbes as well as other organisms; e.g. Dunn and Daniel 2004; Kurkal et al.2005; Lopez et al.2010). It is the norm in biology that individual components of a cellular or organismal system exhibit their optimal performance (and have maximum and minima for function) at different values of water activity and other biophysical parameters (Hallsworth 2018). The increase of hydrogen bonds between protein–surface acidic amino acids and water molecules, and the reduction of hydrophobic patches, enhance both structure and function of enzymes at high salt concentrations (Dym, Mevarech and Sussman 1995; Britton et al.2006; DasSarma and DasSarma 2015). The dynamics of protein folding determined through incoherent neutron spectroscopy of the extreme halophile *Halobacterium salinarum* demonstrated that a large proportion of the proteins in the proteome is strongly dependent on high salt concentrations: cellular vitality becomes compromised at K<sup>+</sup> concentrations below 2.2 M, with many proteins misfolding, precipitating and forming aggregates (Vauclare et al.2015).

In general, the acidity of cellular proteins correlates with the salinity of the environment in which halophiles live. For example, the genomes of the extreme halophiles *Acetohalobium arabaticum* and *S. ruber* exhibit extremely acidic proteomes (DasSarma and DasSarma 2015). In general, acidic proteomes are found in salt-in strategists, i.e. the most-extreme halophiles typically have a more acidic proteome. Moderate halophiles and very halotolerant species (e.g. representatives of Halomonadaceae family) are typically salt-out strategists with a less acidic proteome (Coquelle et al.2010; Schwibbert et al.2011; Oren 2013). For salt-in halophiles, the presence of acidic amino acids at protein surfaces correlates with cellular tolerance to salinity. This enables the use of mutations in genes which encode protein-surface amino acids, resulting in the conversion of a halophilic enzyme into a non-halophilic one, demonstrated empirically with the DNA ligase of Haloferax volcanii (Tadeo et al.2009), Halobacterium diphosphate kinase (Ishibashi et al.2012), ProtL from Peptostreptococcus magnus (Qvist et al. 2012), and more recently in the protein  $\beta$ -trefoil (Longo *et al.***2015**). In conclusion, the ability of halophiles to function in conditions with moderately low water activities (down to ~0.755) is based on the evolution of enzymes

that are stable, and work optimally, in these conditions. This hyperstability has evolved through specific modifications of the surface of the enzymes to establish stable hydrogen bonds with water molecules and reduce the exposed hydrophobic patches (Winter *et al.*2009).

At saturation, NaCl is permissive for the activity of extracellular enzyme systems of extreme halophiles, as evidenced by the functional diversity of enzymes observed in hypersaline environments and the well-characterized adaptations of these proteins (Ferrer et al.2005, 2012; Borin et al.2009; Pachiadaki et al.2014). For instance, an extracellular protease from a saltern bacterium was found to be dependent on NaCl for its functionality, and exhibited optimum catalytic activity at 4.5 M NaCl (Vidyasagar et al.2009). Another protease, from *Haloferax lucentense*, exhibited a similar NaCl optimum, and also retained high activity at NaCl saturation (Manikandan, Pašić and Kannan 2009). *Halobacterium* strains are also known to produce extracellular proteases which function optimally close to NaCl saturation (Norberg and Hofsten 1968), and halophilie communities are a source of industrially important enzymes which are able to function under various extremes (Ferrer et al.2005; Kumar et al.2012; Alsafadi and Paradisi 2013; Moreno et al.2013).

The genes for these highly adapted proteins do not appear to be confined to individual strains and species. Genome sequencing has revealed that high levels of horizontal gene transfer occur between members of natural consortia of halophilic archaea (DeMaere *et al.*2013), and the transfer of large DNA segments (310–530 kb) between different species of *Haloferax* has been observed in laboratory-based experiments (Naor *et al.*2012). Recently, a membrane vesicle-mediated DNA-transfer mechanism has been discovered in *Halorubrum lacusprofundi* (Erdmann *et al.*2017). This means that genes for halophilic proteins that confer an ecological advantage do not need to evolve by the relatively slow process of mutation and selection, *in situ*, but can spread readily between diverse members of a microbial community.

### Other adaptations of halophiles

In addition to osmoadaptive strategy, other survival mechanisms are exhibited by extreme halophiles. For example, sequestration of resources via intracellular storage improves competitive ability and enables tolerance of prolonged periods without carbon substrate, phosphate, etc (Cray *et al.*2013a; Oren and Hallsworth 2014). Some members of the Halobacteria store polyhydroxyalkanoates in the form of intracellular

granules (Saponetti et al. 2011), while Dunaliella species can accumulate triacylglycerols as an overflow product of photosynthesis (Chen et al.2011). Extreme halophiles can also have adaptations which enhance the generation of cell-available energy, such as lightactivated proton pumps, that use light to generate a proton motive force, and have been found in species such as S. ruber and Hqr. walsbyi (Balashov et al.2005; Cray et al.2013a). Some species such as Hqr. walsbyi and Hfx. mediterranei utilise gas vesicles, which facilitate buoyancy and may also maintain the horizontal orientation of the cell thereby optimizing light capture; this may also enhance oxygen acquisition by maintaining the cell's position within the oxygenated regions of stratified brines (Englert, Horne and Pfeifer 1990; Bolhuis et al. 2006; Cray et al. 2013a, Oren and Hallsworth 2014). Halophiles can be exposed to high levels of ultraviolet light, and some species such as D. salina, S. ruber and most haloarchaea synthesise and accumulate high concentrations of carotenoids, thereby protecting their macromolecular systems from ultraviolet radiation (Antón et al.2002; Chen and Jiang 2009). It should be noted that some stresses experienced by, and cellular adaptations of, halophiles are not unique to these microbes. Indeed, cellular stress is the norm for all microbial systems and many sources of stress impact the cell via via the same mechanism, regardless of microbial species or habitat type (Hallsworth 2018).

Recently, a number of studies have utilised (culture-independent) analytical techniques to identify metabolites of halophilic microbes, including mass spectrometry coupled with gas or liquid chromatography and/or capillary electrophoresis (e.g. Kido Soule *et al.*2015; Johnson, Kido Soule and Kujawinski 2017). The comprehensive power of these technologies should in future reveal as-yet-unidentified compounds or strategies underlying the cellular phenotype of extreme halophiles.

# EVOLUTIONARY BIOLOGY OF HALOPHILES; IMPLICATIONS OF THE FINITE SOLUBILITY OF \$NaCl\$

For all types of environments (saline or not), habitability by microbes is constrained or prevented by one or more factors; these are usually nutritional, physicochemical and/or biophysical. Environmental parameters—including temperature, water activity, irradiance, ultraviolet radiation, pH, chaotropicity and availability of electron donors or nitrogen substrates—determine biodiversity, abundance, overall biomass and ecosystem function. To put it simply, ecosystems (and the biosphere) are shaped and

restricted by available resources and prevailing conditions. And yet, there are few microbial ecosystems in aquatic environments which appear to outperform NaCl-rich microbial habitats. Just as tropical rainforest is considered the pinnacle of biomass and biodiversity when compared with the other plant biomes (Pan et al.2013), the halophile communities of solar salterns and deep-sea brines may represent the greatest attainment of aquatic ecosystems in relation to overall metabolism and cell densities. By contrast, in ethanol-producing fermentations, the microbial community is biomass - and diversity limited due to the chaotropicity of ethanol, low pH and low water activity (Cray et al.2015). Whereas NaCl-saturated environments clearly select for halophiles, the growth rates, cell densities and microbial abundance of these inhabitants are remarkably high (Table 1, Fig. 2A-E). This is because the NaCl concentration limit is not determined by biology (as discussed above, 5 M NaCl is not the theoretical maximum concentration at which the cellular systems of extreme halophiles can function), but by a quirk of chemistry. The dipoles of water molecules attract ions more than Na and Cl ions attract each other, and these ions (especially Na ions) attract water more than water molecules attract each other, hence the kosmotropicity of NaCl (Cray et al.2013b; Alves et al.2015). At the saturation point, however, most of the water is interacting with these ions so it becomes more thermodynamically favourable for additional ions to bond with each other; i.e. precipitation of NaCl occurs (Fig. 2f).

The empirically determined and theoretical water–activity limits for fungal cell division and those for growth limits for halophilic prokaryotes (0.585 and 0.565, and 0.635 and 0.611, respectively) are equivalent to a theoretical NaCl concentration of up to 8 M (equivalent to 47%, w/v). This is consistent with the thesis that 5 M NaCl provides mid–range (i.e. moderate) conditions for numerous halophiles (Fig. 2A–E; Stevenson *et al.*2015b). Whereas specific temperatures or ethanol– or MgCl<sub>2</sub> concentrations act as barriers to life, there is no threshold value at which NaCl prevents microbial habitation (Figs 1A, B and 2). At NaCl saturation, therefore, there is considerable thermodynamic leeway for extreme halophiles before the theoretical failure of the cellular system is reached.

Whereas the finite solubility of NaCl does not curtail ecosystem development, the biology of some saturated brines may be constrained by the interactive effects of other limiting factors. These include nutrient deficiency (e.g. scarcity of phosphate in hypersaline brines can restrict the growth of phototrophs such as *Dunaliella*; Gasol *et al.*2004; Oren 2017); high pressure and extremes of pH or temperature (Harrison

et al.2013); high irradiance (Wieland and Kühl 2000); or high concentrations of chaotropic ions (see above) or toxic metals, such as arsenic, though there are some halophile communities that function in arsenic-containing alkaline brines (Oremland and Stolz 2003; Oremland et al. 2005). Populations of Bacteria and Archaea can be constrained by the activities of viruses and/or grazers; for instance, bacteriophageinduced lysis and nanoflagellate grazing have been identified as key factors reducing prokaryote abundance in Spanish crystalliser ponds and Korean salterns, respectively (Guixa-Boixareu *et al.*1996; Park *et al.*2003; Cray *et al.*2013a). The ecological importance of viruses within halophile communities was highlighted in a study which successfully assembled several viral genomes from the metagenome of crystalliser pond CR30 of Santa Pola salterns (Garcia-Heredia et al. 2012). On the basis of shared CRISPR spacer sequences, some of these viruses were found to prey upon *Hqr. walsbyi*. Those salterns with 3.6-5 M NaCl were populated primarily by members of the Euryarchaeota or Nanohaloarchaeota (making up to 80-90% of the community and mostly represented by Hgr. walsbyi and Nanohaloarchaea) and Bacteroidetes (9–10%; mainly Salinibacter ruber) (Garcia-Heredia et al.2012), so their phage associations further underscored that viruses are ubiquitous within, and central to the ecologies of, NaCl-saturated habitats.

Novel halophilic methanogenic archaea have been isolated from the anaerobic sediments of alkaline hypersaline lakes, soda lakes and pH-neutral hypersaline lakes (Sorokin et al. 2017a). The candidate 'Methanohaloarchaea' (HMET) and 'Methanonatronoarchaea' (AMET) strains were capable of methylotrophic methanogenesis at slightly thermophilic  $(50-60^{\circ}\text{C})$  and either saturated NaCl (pH 7) or soda (pH 9.6) conditions, respectively. These strains belong to 'Methanonatronoarchaeia' (former SA1 clade), a distinct class-level deep phylogenetic group within Euryarchaeota possessing the salt-in strategy. Further phylogenomic analyses have suggested that a common halophilic, methanogenic ancestor might have diverged into 'Methanohaloarchaeia' (by gene loss; Spang, Caceres and Ettema 2017; Sorokin et al. 2017a) and Halobacteria (by massive interdomain gene transfer from aerobic bacteria; Nelson-Sathi et al. 2012, 2015; Sousa, Nelson-Sathi and Martin 2016). Although the Halobacteria appear to be comprised primarily of aerobic and chemoorganoheterotrophs with a halophilic lifestyle, a few are facultatively anaerobic (Hattori et al.2016) or fermenting (Antunes et al.2008). However, recent studies allude to greater metabolic capabilities within this group. Halanaeroarchaeum sulphurireducens, for example, is a strictly anaerobic, sulphidogenic haloarchaeon that utilises acetate and pyruvate while reducing elemental sulphur (Sorokin et al.2016b, c), whereas Halodesulphurarchaeum formicicum is the first evidenced lithotrophic

Halobacteria member that uses  $H_2$  or formate as an electron donor and reduced sulphur compounds (thiosulphate, dimethylsufides and  $S^0$ ) as electron acceptors (Sorokin *et al.*2017b).

The finite solubility of NaCl raises a number of questions in relation to the evolutionary biology of halophiles. In particular, it remains to be investigated how evolvable halophiles are in their natural environments. Because of the strong selective constraints imposed by the requirement for highly acidic proteins, the fixation rate of mutations in halophiles is expected to be low. In fact, the mutation rate of some halophiles, such as *Hfx. volcanii*, is lower than that of microbes that are not halotolerant or halophilic (Mackwan et al. 2007; Busch and DiRuggiero 2010). Since the fixation rate of mutations is much lower than the mutation rate, it is reasonable to assume a lower fixation rate of mutations for halophiles than for most other microbes. Mutation rates in thermophiles are also reported to be lower than in those for mesophiles (Friedman et al.2004; Kissling, Grogan and Drake. 2013), and it has similarly been argued that the functional constraints are much higher for thermophilic than for mesophilic proteins. The greater-than-expected genetic diversity seen in Halobacteria, such as the genera Halorubrum and Haloarcula, is produced mainly by horizontal gene transfer and homologous recombination (Naor et al. 2012; DeMaere et al. 2013; Mohan et al. 2014). This genome-level diversity could in principle generate a reservoir of genomic variants that can be maintained phenotypically cryptic in the population, but with enormous potential to originate adaptations to other environmental perturbations. In NaClsaturated environments, selection becomes stronger against many halotolerant (and some halophilic) microbes allowing the persistence of only those genetic variants able to function optimally at 0.755 water activity. However, NaCl is not sufficiently soluble to reduce water-activity values to the 0.755-0.585 range experienced by some extremophiles. Those salts which are sufficiently soluble (e.g. MgCl<sub>2</sub>) are so chaotropic that the cellular systems of most halophiles are inhibited by their chaotropicity rather than reduced water activity. If genetic variants arose that could grow at water activities below 0.755, even to 0.585, they are not likely to be any fitter than existing species in natural hypersaline environments, and without positive selection are unlikely to maintain or increase their frequency. At the water-activity limit for life, molecular crowding within the cytoplasm and/or inadequate hydration or flexibility of macromolecular systems may act to prevent cellular function (Fields 2001; Zaccai 2004; Miermont *et al.*2013; Parry *et al.*2014). It has yet to be demonstrated that such factors are indeed limiting in NaCl-saturated brines at 0.755 water activity.

## IMPLICATIONS FOR ASTROBIOLOGY AND CONCLUDING REMARKS

There are very specific concentrations/proportions of NaCl-MgCl<sub>2</sub> or NaCl-MgCl<sub>2</sub>-glycerol mixtures that permit the proliferation of halophiles below 0.755 water activity (Stevenson *et al.*2015a,b). The highest concentrations of NaCl that can be experienced by the cells of halophiles, 5 M, enables optimum rates of metabolism and growth for some populations, and facilitates the development of biomass dense- and highly active ecosystems. Therefore, we conclude that the environments with NaCl concentrations of ~5 M are not thermodynamically limiting and, thus, this concentration does not represent a biophysical limit for life on Earth. Given the high cell densities and broad functional diversity of many hypersaline brine systems, it is intriguing to ask why their microbial diversity is not even higher. Can this be explained as a product of dominance by a small number of halophile weed species, or are there other factors at play? The solution chemistry of NaCl raises additional questions pertinent to the field of astrobiology:

- i. Why are there no truly psychrophilic haloarchaea?
- ii. Whereas chaotropic brines may be more readily habitable at subzero temperatures (Ball and Hallsworth 2015), would the evolutionary trajectories of terrestrial halophiles, which are not usually exposed to habitable brines with water activities much below 0.755, make them unable to function in chaotropic brines on Mars or Europa (Hallsworth *et al.*2007; Chin *et al.*2010; Rummel *et al.*2014)?
- iii. Would the diverse solution chemistries found on different planetary bodies (e.g. Muñoz-Iglesias, Bonales and Prieto-Ballesteros 2013; Fox-Powell *et al.*2016) result in qualitative differences in the evolution of terrestrial halophiles and their ability to tolerate multiple extremes not found on Earth, and could terrestrial, extreme halophiles tolerate the high ionic strength which characterises some martian brines (Fox-Powell *et al.*2016)?
- iv. Diverse NaCl stoichiometries have been reported at high pressures and temperatures (i.e. 200 000 atmospheres and 2000 K), resulting in the formation of stable salts such as NaCl<sub>3</sub> (Zhang *et al.*2013). Such salts may be formed during accretion and deep within a planetary or moon body, but only small amounts would be present at

or near the surface. Whereas, they may not exist at high concentrations under conditions which are pertinent to (known forms of) life, this finding does raise the question: Do sodium chlorides, that are more soluble than NaCl, reduce water activity to a lower value and exhibit chaotropic activity?

v. Given that saturated NaCl is not stressful for some halophile systems, would these species act as pioneers in the event that NaCl-rich sites on other planetary bodies are contaminated during space exploration missions?

The evolution of life on early Earth is thought to have taken place in extreme aqueous environments such as hydrothermal vents and acid-brine oceans. Hypersaline brines have remained a stable habitat throughout the planet's geological history (Warren 2010), and Buick and Dunlop (1990) describe evaporite deposits from 3.5 Ga Warrawoona Group of Western Australia, confirming the presence of brine surface environments at the time of some of the first life on Earth. Diverse lines of evidence suggest that terrestrial life may have inhabited, and could have even arisen in, brines with a high NaCl, KCl and/or MgCl<sub>2</sub> concentration (e.g. Dundas 1998; Stevenson *et al.*2015a; Matveev 2017). Water activity and chaotropicity are likely to have been key determinants for both habitability and the evolution of early cells in relation to early life on Earth (Stevenson *et al.*2015a).

The likelihood that early life was energised by electrochemical  $Na^+$  gradients has been discussed by Price *et al.* (2017), who revealed steep natural gradients (i.e. 4-52 mM  $Na^+$  in the venting liquids versus 468 mM  $Na^+$  in the surrounding sea water) across thin mineral layers, figuratively termed 'membranes', within the pores of serpentinite located in shallow sea water. The cell-sized serpentinite pores percolated by the alkaline, reducing ( $H_2-$  and  $CH_4-$ rich) fluids which contain  $Na^+$  at concentrations one to two orders of magnitude less than in sea water, may have acted as incubators for the development of protocells on the bottom of the Hadean oceans (Sojo *et al.*2016; Price *et al.*2017).

In relation to astrobiology, the occurrence of halophily throughout the phylogenetic tree suggests that it is relatively easy to evolve, and that it evolved more than once. If life evolved in locations beyond Earth, and where salty brines occur, then there would almost certainly be the evolution of halophilic organisms (Mancinelli 2005a,b). There is strong evidence for the past existence of acid brines on Mars (Squyres *et al.*2004; Benison and Bowen 2006). Mars hosts abundant saline minerals, including chlorides

(likely halite) and hydrated calcium sulphates (likely gypsum) in lake deposits and lake-derived eolian deposits (Glotch *et al.*2010). The presumed acid- and brine nature of past Mars surface waters have led to speculation that Mars would be inhospitable to life (Tosca, Knoll and McLennan 2008; Rummel *et al.*2014). However, in light of the findings of this study, halophiles may potentially be able to evolve the ability to proliferate at, or below, 0.585 water activity. Therefore, the possibility that such microbes can potentially contaminate saline milieus present on Mars in this water activity range should not be discounted. The findings reported in the current article also expose a need for new experimental and theoretical approaches, which circumvent the apparent limits for life which are imposed merely by solution chemistry, to identify and characterise the thermodynamic constraints of halophilic life.

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